

Effects of Rapid Thermal Annealing on the Properties of Room-Temperature Oxygenated DC Sputtered Zinc Thin Films for CZTS Solar Cells Application

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Abstract

This work investigated the potential to achieve zinc oxide (ZnO) films for Cu_2ZnSnS_4 (CZTS) solar cells window layer at controlled annealing conditions as a potential approach to address elemental inter-diffusion in CZTS solar cells. This involved rapid thermal annealing (RTA) of room-temperature oxygenated DC sputtered zinc thin films in an ambient of nitrogen gas at different temperatures. Structural, morphological, optical, and electrical properties of these films were determined by X-ray diffractometer, Scanning Electron Microscopy, Ultraviolet-visible-near infrared spectrophotometer, and Hall Effect measurement, respectively. ZnO phases were observed after annealing the films over 150 °C. The films' grains sizes improved with increasing RTA temperature attaining the lowest value at 300 °C. The bandgap and average solar transmittance of the films increased with increasing RTA temperature achieving values that are potential for applications in CZTS solar cells window layer at RTA temperatures beyond 200 °C.

Keywords: Sputtering; Rapid thermal annealing; Zinc oxide; Structural; Opt-electrical.

Introduction

Zinc oxide (ZnO) thin films have been broadly researched due to their potential properties applicable in different fields. These properties include the wide tunable bandgap (3.0-3.4 eV) and high exciton binding energy (60 meV). Other properties include low-cost, non-toxicity, and thermal and chemical stability (Pholds et al. 2017). Therefore, ZnO has been researched for sensors, acoustic waves devices and optoelectronic devices such as light-emitting diodes and solar cells (Ozgur et al. 2010, Samwel et al. 2015). These films have been achieved using several methods including chemical vapour deposition, pulsed laser deposition, spray pyrolysis, molecular beam epitaxy, spin coating, thermal evaporation and sputtering. Among these methods, sputtering which is also reported to be versatile in the ZnO thin films production, achieve high-quality uniform films with exceptional physical and structural properties (Zhang et al. 2009).

ZnO thin films are usually sputtered at elevated substrate temperature to reduce structural defects, improve grains growth, and achieve crystalline films (Jang et al. 2010). However, these films have also been produced at low-temperature followed by controlled post-deposition heat treatment including rapid thermal annealing (RTA). Such an annealing approach is reported to be essential in achieving defects-rich ZnO films for specific applications (Hsieh et al. 2006). The lowtemperature deposition has also been used in ZnO thin films to avoid cracking of the films due to high thermal expansion between substrate and films interface (Barker et al. 1997) and/or avoid adverse effects of underlayers or substrates (Ryu et al. 2017). In Cu₂ZnSnS₄ (CZTS) solar cells processing, the ZnO window layer is normally deposited on a CdS buffer layer at suitable temperatures to achieve a good quality window layer (Husna et al. 2012). These deposition temperatures can also be considered as an annealing process on the CdS under-layer usually deposited at relative lower room temperatures. Such temperatures have been reported to cause cadmium (Cd) inter-diffusion into CZTS and Zn and S into CdS affecting CdS/CZTS band alignment which disrupt carrier transport at this interface eventually affecting the CZTS solar cells performance (Pal et al. 2019). A potential approach to address this reported issue is to achieve ZnO thin films on CdS at controlled annealing environment. In this work, we assessed the potential to achieve ZnO thin films by RTA of room-temperature oxygenated DC sputtered Zn thin films in an ambient of nitrogen gas at different temperatures; an approach which is considered to be potential to address the aforementioned adverse effect in CZTS solar cells related to deposition temperature. ZnO films achieved by DC sputtering of Zn metallic target provide much more control of the films' oxygen content which is potential to tailor ZnO films properties, whereas RTA in nitrogen gas ambient helps to exclusively understand the influence of the oxygen content contained in these Zn metallic films under this controlled annealing processes.

Materials and Methods

Zinc thin films were DC sputtered on glass substrates by magnetron sputtering of high purity (99.99%) Plasmaterials Inc. metallic disk (diameter 5.08 cm) zinc target at room temperature in argon-oxygen gas environment followed by post-deposition RTA at different temperatures. The sputtering chamber was evacuated to a base pressure of 5.0 \times 10⁻⁶ mbar before films deposition. High purity argon gas (99.99%) was allowed into the chamber at 50 ml/min. Pre-sputtering was carried out for 5 minutes to remove contaminants from the surface of the target. Oxygen gas of high purity (99.99%) was allowed into the chamber at 50 ml/min. Films of 300 nm were deposited for 10 minutes on the substrates located about 15 cm from the target at a working pressure of 6.0×10^{-3} mbar and sputtering power of 175 W. Temperature change on the surface of the substrate during sputtering was monitored using a thermocouple embedded on the substrate surface as shown in Figure 1.

The films were RTA at different temperatures in an evacuated rapid thermal processing (RTP) system chamber containing nitrogen gas. The as-deposited films were loaded into the RTP system chamber which was evacuated using the rotary pump for 30 minutes. Nitrogen gas (purity 99.99%) was allowed into the chamber at 60 ml/min through a gas controller. The RTP was ramped up at 20 °C/sec to the desired temperature. The films were annealed for 5 minutes at 100 °C, 150, °C, 200 °C, 250 °C and 300 °C. After annealing the system were allowed to cool down naturally to room temperature. The annealed films were analysed for structural, morphological, optical, and electrical properties. Films' thicknesses were determined using the Alpha Step surface profiler. The films' surface morphology was determined using Scanning Electron Microscopy (SEM) operating at 2.0 kV. The films' grains distribution and surface roughness were analysed from the SEM images using Image J software (v1.53g60) (Rueden et al. 2017) and WSxM software (version 4.0) (Horcas et al. 2007), respectively. The structural properties were determined using an X-ray Diffractometer (XRD) using 0.1504 nm wavelength Cu Kalpha radiation. The reference data from the RUFF project database (Lafuente et al. 2015) were used to determine the phases of the films. The transmittance spectra (T_{sol}) in the

wavelength range 250-2500 nm were determined using UV/VIS/NIR spectrophotometer (Perkin Elmer Lambda 19). The electrical resistivity, and carrier concentrations and mobility were determined using the Hall Effect Measurement system (Ecopia HMS 3000).



Figure 1: Schematic diagram of the sputtering chamber with a thermocouple connecting the substrate surface to the outside digital multi-meter through the chamber's feed-through.

Results and Discussion XRD analysis

The XRD patterns (Figure 2a) showed for the as-deposited films intermediate peaks between Zn (RRUFF ID: R070497) and ZnO (RRUFF ID: R050492) phases suggesting that mixed solid-phase of these materials form at room temperature. XRD studies on similar films sputtered at room temperature reported amorphous (Barker et al. 1997) or ZnO phase (Zhang et al. 2009). The mixed-phase observed in this work is likely driven by energy supplied by sputtered species (Georgieva et al. 2009) related to sputtering parameters. Films that were annealed at 100 °C revealed no structural change relative to the as-deposited films implying that this annealing temperature is still insufficient to form distinct ZnO phases. However, at an annealing temperature of 150 °C distinct Zn and ZnO peaks were observed. For temperatures above that, the peaks still

existed with some variations above or below that at 150 °C. Peaks observed at around 34.4° were assigned to (002) planes of ZnO (RRUFF ID: R050492), while other peaks were assigned to Zn (RRUFF ID: R070497) and SiO₂ (RRUFF ID: R060648). The co-existed Zn and ZnO phase can be ascribed to saturation of ZnO formation likely due to depleted oxygen contained in the as-deposited films. The silicon oxide phase observed likely originates from the glass substrate and have also been reported for room-temperature deposited ZnO (García-Gancedo et al. 2012). These findings suggest that the present deposition approach is potential for ZnO thin films phases and structural control that might be of interest for applications in optoelectronics (Reshchikov et al. 2007).

To assess the growth of the ZnO phases observed with annealing temperature, we computed the crystallite size related to (002) planes assigned to ZnO thin films using the Debye-Scherer formula

 $D = 0.94\lambda / \beta \cos\theta \tag{1}$

where, D, λ , θ , and β represent grain size, XRD radiation wavelength, Bragg angle, and Full-Width at Half Maximum, respectively. The analysis by this formula revealed a general increase in ZnO crystallite sizes with increasing annealing temperature (Figure 2b). The relative lower ZnO crystallite size for films that were annealed at 200 °C as compared to the rest of the films might be related to a release of adsorbed oxygen gas from the film's surface at this annealing temperature (Mahmood et al. 2010).



Figure 2: XRD pattern (a), and the crystallite size (b) for the as-deposited and annealed ZnO thin films.

SEM analysis

The SEM images (Figure 3a) showed evenly distributed large and small roundish clustered grains for the as-deposited films suggesting films' grain grows before annealing processing. Such growth of grains before annealing processing has been reported for ZnO thin films deposited at room temperature (Jang et al. 2010). This has been mostly ascribed to un-intentional substrate heating resulting from energetic atoms bombarding the surface of this substrate during films sputtering (Nennewitz et al. 1994). Increasing temperature due to these bombarding atoms is an unlikely cause of the films' grains observed here prior to annealing as the change in the substrate temperature during sputtering was negligibly small (from 25 °C to 29 °C) to cause these films' observed grains growth. The small

change in substrate temperature is likely due to the relatively large target to substrate distance and short sputtering time used in this work. The films' grains growth around room temperature can be explained by energy provided by the depositing species (Georgieva et al. 2009) rather than the elevated temperature of the substrate. Relative smaller and well-defined grains are observed for films that were annealed at 300 °C (Figure 3b) as compared to the as-deposited films (Figure 3a) suggesting an evolution of the films surface annealing morphology with increasing temperatures (Jang et al. 2010). Findings from this analysis suggest that annealing of films deposited using this approach is potential for these thin films surface morphology engineering.



Figure 3: SEM images and grains distribution for as-deposited (a, c) and annealed at 300 °C (b, c) ZnO thin films.

analysis of the films' The grains distribution revealed nearly normal distributed grains for both the as-deposited (Figure 3c) and films that were annealed at 300 °C (Figure 3d) with average grain sizes of 45.2 nm and 29.8 nm, respectively (Table 1). This showed relative decrease in grains sizes which can be explained by an increase in films' crystallization with an increase in annealing temperature (Hong and Ro 2013). The larger grain size for the as-deposited films as compared to its crystallite size (Table 1) suggests that the estimated grain size included clustered grains unlike crystallite size determination which depended on localized ZnO phase crystallites. The grains get less clustered and become nearly spherical in shapes for the films that were annealed at 300 °C (Hezam et al. 2010). Therefore, achieving relatively similar grain sizes determined from SEM lateral surface scanning (29.8 nm) and the crystallite size determined from XRD vertical surface probing (31.5 nm). Relative higher surface roughness and root-mean-square roughness (RMS) were observed for films that were annealed at 300 °C as compared to the asdeposited films (Table 1). This suggested an increase in surface roughness with the films' oxide formation as phase annealing temperature increases (Daniel et al. 2010). These grains sizes and roughness parameters are within the reported values potential for optoelectronics applications (Malik et al. 2019).

 Table 1: Crystallite size, grain size, and surface parameters for the as-deposited and annealed at 300 °C ZnO thin films

Parameter	As-deposited	RTA at 300 °C
Crystallite size (nm)	15.4	31.5
Grain size (nm)	45.2	29.8
Surface roughness, S_a (nm)	0.236	0.403
RMS roughness, S_q (nm)	0.732	0.829

Transmittance properties

Figure 4(a) shows solar transmittance (T_{sol}) spectra that are relatively lower and of different shapes for the samples as compared to that of bare glass substrate an indication that samples of different optical properties were deposited on the bare glass. The area under $T_{\rm sol}$ spectra increased for annealed films as compared to the as-deposited films implying a relative improved $T_{\rm sol}$ due to improved phase content and crystallization (Eneku et al. 2018, Malik et al. 2019) with increasing annealing temperature evident in our XRD data. A relatively large difference in the areas under T_{sol} spectra is observed between the as-deposited, films annealed at 100 °C and 150 °C as compared to that between films annealed at 200 °C, 250 °C and 300 °C with a relatively similar area under $T_{\rm sol}$ spectra. The former suggests that the films undergoing significant phase are and crystallinity evolution resulting in varied $T_{\rm sol}$, while the later signify stable ZnO phase and crystallinity with high and nearly similar $T_{\rm sol}$. Films that were annealed at 250 °C and 300 °C have $T_{\rm sol}$ very close to that of bare glass around the verge of visible-infrared region suggesting

that these films show anti-reflection properties (Granqvist 2007).

The films' average solar transmittance, T_{AVE} was computed based on Equation (2) (Maghanga et al. 2010) using Air Mass 1.5 solar irradiance, $G(\lambda)$ in the range 250–2500 nm.

$$T_{\rm AVE} = \sum_{250}^{2500} T(\lambda) G(\lambda) / \sum_{250}^{2500} G(\lambda)$$
(2)

The computed average transmittance (T_{AVE}) as a function of annealing temperature (Figure 4b) shows a relative low T_{AVE} value for asdeposited as compared to the plain glass substrate and a general increase of this average with increasing annealing temperature as compared to the as-deposited films. This agrees well with the trend observed for these films T_{sol} spectra (Figure 4a). Films that were annealed at 200 °C and above showed relative high similar values of T_{AVE} at about 80% which is reportedly suitable for applications in thin film solar cells window layer (Khan et al. 2017).



Figure 4: Solar transmittance spectra, T_{sol} (a), and average transmittance (T_{AVE}) (b) as a function of annealing temperature for as-deposited and annealed ZnO thin films.

Bandgap properties

The films' bandgap energy E_g was estimated from the films' absorption coefficient α computed from the solar transmittance spectra $T(\lambda)$ which relate as (Song et al. 2011)

$$T(\lambda) = A \exp(\alpha t) \tag{3}$$

Besides, for ZnO thin films with a direct interband transition, the bandgap and α relate as

$$(\alpha h v)^2 = \beta (h v - E_g) \tag{4}$$





Figure 5: Tauc plots and extrapolations (a) and estimated bandgap (b) for as-deposited and annealed ZnO thin films.

The bandgap increased with increasing annealing temperature up to 200 °C, thereafter recording a relative similar value beyond this temperature. This finding agrees well with the observed $T_{\rm sol}$ spectra properties of these films and can be explained by the similar evolution of microstructure pointed out earlier in the structure analysis. This increase in bandgap is reported to be important for optoelectronic applications (Malik et al. 2019). A relative small variation in the bandgaps for films annealed at 200 °C and beyond is likely due to slightly varied films' microstructure defects and carrier concentrations due to decreased oxygen vacancies with these films annealing temperatures (Dhara and Giri 2011). The bandgap for the best films in the present study is close (the highest achieved is 3.30 eV) to the reported bandgap of ZnO thin films (3.37 eV) (Ghosh et al. 2015). These findings suggest that the fabrication approach in this work is potential to achieve ZnO with bandgap potential for application in CZTS solar cells window layer.

Electrical properties

The Hall Effect measurement showed rather similar carrier concentration values for the asdeposited and films that were annealed at 100 $^{\circ}$ C (Figure 6a). However, films annealed at 150 $^{\circ}$ C and 200 $^{\circ}$ C had relative higher carrier concentrations as compared to the rest of the films. This finding suggests that carriers in the former two films are trapped by stilldeveloping films grains within these films. Relative higher values of carrier concentrations for films that were annealed at 150 °C and 200 °C suggest relative reduced grains carrier trapping with increasing annealing temperature. Further increase in annealing temperature beyond 200 °C revealed a decrease in carrier concentrations likely due to reduced oxygen vacancies responsible for carriers in ZnO films (Ghosh et al. 2015). These films also showed high carrier mobility (Figure 6b) which is likely due to improved crystallinity and reduced carrier-carrier scattering related to these films reduced carrier concentrations (McLean and Paige 1960). Besides, relatively low carrier mobility recorded for the films that were annealed at 150 °C and 200 °C can be explained by increased carrier-carrier scattering due to the increased carrier concentrations in these films. Whereas relative lower carrier mobility for the as-deposited and annealed at 100 °C suggest scattering of the carrier on the still-developing grains at these conditions. The Hall measurement also showed that with increasing annealing temperature, the films' electrical resistivity decreased exponentially (Figure 6c). This decrease can be explained by the combined effects of these films' carriers' concentrations and mobility related to these films microstructures.



Figure 6: Carrier concentrations (a), carrier mobility (b), and resistivity (c) for as-deposited and annealed ZnO thin films.

Conclusion

In this work we found that, potential ZnO thin films desired for CZTS solar cells window layer applications can be achieved by RTA of room-temperature oxygenated DC sputtered Zn thin films carried out under nitrogen gas environment. The observed desired properties as window layer applications include the relative high films grains, optical bandgap and average solar transmittance. ZnO phase was also observed for films RTA at 150 °C and above with an impact on the electrical properties of the films indicating a potential to grow ZnO thin films purity phases at these annealing temperatures upon further sputtering parameter optimization. The best films had a high T_{AVE} of about 80%, E_{g} of 3.30 eV and electrical resistivity of $1.74 \times 10^3 \Omega$ cm achieved at an annealing temperature of 300 °C. The present deposition approach extends the possibility to achieve ZnO thin films for the CZTS thin films solar cells window layer.

Conflict of Interest

There is no conflict of interest to declare.

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